


# Safety Data Sheet

Section 1. Identification	
<b>Product Identifier</b>	<b>Sulphamic Acid</b>
	<b>Version: 6</b> <b>Effective Date: 25 January 2021</b>
<b>Other Means Of Identification</b>	Amidosulfonic Acid
<b>Initial Supplier Identifier</b>	Chemfax Products Ltd. 11444 – 42 Street SE Calgary, AB T2C 5C4 Tel: 403-287-2055
<b>Recommended Use and Restrictions On Use</b>	Industrial acid, reactant, chemical intermediate, descaler. No restrictions.
<b>Product Family</b>	Organic acids
<b>Emergency Phone</b>	1-855-887-2055 Monday - Friday 8:00am - 4:30pm MST

Section 2. Hazard Identification	
<b>Hazard Classification</b>	
<b>Health Hazards</b>	Skin Corrosion/Irritation – Category 2 Eye Damage/Irritation – Category 2A
<b>Environmental Hazards</b>	Hazardous To The Aquatic Environment – Short Term (Acute) Hazard – Category 3 Hazardous To The Aquatic Environment – Long Term (Chronic) Hazard – Category 3
<b>Signal Word</b>	Warning
<b>Hazard Statement</b>	Causes skin irritation. Causes serious eye irritation. Harmful to aquatic life with long lasting effects.
<b>Precautionary Prevention Statement</b>	Wash hands thoroughly after handling. Wear protective gloves, eye and face protection. Avoid release to the environment.
<b>Precautionary Response Statement</b>	IF ON SKIN: Wash with plenty water. Specific Treatment: Treat symptomatically. Do not induce vomiting unless directed by medical personnel.

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	<p>If skin irritation occurs: Seek medical attention. Remove contaminated clothing and wash it before reuse.          IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. If eye irritation persists, seek medical attention.</p>
<b>Precautionary Storage Statement</b>	No statement
<b>Precautionary Disposal Statement</b>	Dispose of contents/container in accordance with local regulations.
<b>Other Hazards</b>	None

### Section 3. Composition / Information on Ingredients

Chemical Name	Common Name or Synonyms	CAS NO. and Other Unique Identifiers	% by weight
Sulphamic acid	Amidosulphonic acid	5329-14-6	99

### Section 4. First-Aid Measures

<b>Eye Contact</b>	Flush eyes with water for 15 minutes. Seek medical attention.
<b>Skin Contact</b>	Flush area with water. If irritation persists, seek medical attention. Launder clothing before reuse.
<b>Inhalation</b>	Remove victim to fresh air. If there is difficulty breathing, seek immediate medical attention.
<b>Ingestion</b>	Remove victim to fresh air. If there is difficulty breathing, seek immediate medical attention.
<b>Most Important Symptoms and Effects Both Acute and Delayed</b>	Irritation, headache, shortness of breath, and nausea.
<b>Immediate Medical Attention and Special Treatment</b>	No specific treatment. Treat symptomatically.

### Section 5. Fire-Fighting Measures

<b>Suitable and Unsuitable Extinguishing Media</b>	Dry chemical, CO <sub>2</sub> , alcohol foam, or water spray.
<b>Hazardous Combustion Products</b>	Oxides of nitrogen, oxides of sulphur, oxides of carbon, and ammonia.



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<b>Specific Hazards Arising From the Product</b>	Thermal degradation may produce toxic gases and mist, such as sulphur oxides and nitrogen oxides.
<b>Special Protective Equipment and Precautions For Fire-Fighters</b>	Fire-fighters should wear self-contained breathing apparatus and full protective clothing. Use water spray to cool containers and structures exposed to fire.

## Section 6. Accidental Release Measures

<b>Personal Precautions, Protective Equipment and Emergency Procedures</b>	Chemical resistant gloves (neoprene), coveralls (rubber) and boots rubber. Isolate area and evacuate unnecessary personnel. Ensure adequate ventilation.
<b>Environmental Precautions</b>	Do not allow to enter the storm water systems or surface drains. Dike and contain.
<b>Methods and Materials For Containment and Clean-Up</b>	Sweep up spilled product without generating dust. Place in a suitable container for disposal. Neutralize with lime slurry or soda ash. Flush area with water to remove residues.

## Section 7. Handling and Storage

<b>Precautions For Safe Handling</b>	Handle with care. Corrosive in contact with moisture.
<b>Conditions For Safe Storage</b>	Store in a cool dry place. Keep containers closed when not in use. Protect against water, moisture, and physical damage. Avoid incompatible materials.

## Section 8. Exposure Controls / Personal Protection

<b>Control Parameters</b> Sulphamic acid	<b>TWA: 8 Hr</b> <b>STEL: 15 min</b> <b>Ceiling</b> <b>IDLH *</b> No information available
	* Immediately Dangerous to Life and Health
<b>Exposure Controls</b>	Local exhaust ventilation
<b>Appropriate Engineering Controls</b>	Provide exhaust ventilation to keep the airborne concentration below the applicable workplace exposure limits. Ensure safety shower and eyewash station are available.
<b>Individual Protective Measures</b>	For dusty or misty conditions:
<b>Eye / Face Protection</b>	Safety glasses.
<b>Skin Protection</b>	Chemical resistant coveralls, gloves
<b>Respiratory Protection</b>	NIOSH approved respirator

<b>Section 9. Physical and Chemical Properties</b>	
<b>Appearance</b>	White crystals
<b>Odour</b>	No odour
<b>Odour Threshold</b>	Not available.
<b>pH</b>	1.18
<b>Flash Point</b>	Not applicable
<b>Boiling Point and Boiling Range</b>	Decomposes at 209 °C
<b>Melting Point and Freezing Point</b>	205 °C
<b>Evaporation Rate</b>	Not applicable
<b>Flammability (solid, gas)</b>	Not flammable
<b>Upper and Lower Flammability or Explosive Limits</b>	No data
<b>Vapour Pressure</b>	Not applicable
<b>Vapour Density</b>	Not applicable
<b>Relative Density</b>	2.125
<b>Solubility</b>	14.7 % @ 0 °C
<b>Partition co-efficient, n-Octanol/Water</b>	No data
<b>Auto-ignition Temperature</b>	No data
<b>Decomposition Temperature</b>	209 °C
<b>Viscosity</b>	Not applicable

<b>Section 10. Stability and Reactivity</b>	
<b>Reactivity</b>	Stable
<b>Chemical Stability</b>	Stable when dry, but hydrolyses in solution. Sulphamic acid decomposes at 208 °C. At room temperature, dilute solutions are stable for many months. At higher temperatures and with stronger solutions, hydrolysis of the acid and its ammonium salts occurs, forming ammonium hydrogen sulphate and ammonium sulphate. This reaction occurs much more rapidly as the pH lowers.
<b>Possibility of Hazardous Reactions</b>	Will not occur
<b>Conditions to Avoid</b>	Avoid dispersions of sulphamic acid particulates into air and contact with heat. Avoid the use of non vented containers if concentrated solutions of the acid are made and heated, as a runaway hydrolysis reaction will occur, generating sufficient steam to cause an explosion.
<b>Incompatible Materials</b>	Chlorine, potassium chlorate, nitric acid, metal sulphides, nitrates, nitrites, strong bases, bromine, and strong oxidizers.
<b>Hazardous Decomposition Products</b>	Thermal decomposition products may include toxic and corrosive fumes of ammonia, and toxic oxides of nitrogen and sulphur. NOTE:

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Chlorination of sulphamic acid with acidic ammonium chloride solutions gives the powerfully explosive oil, nitrogen trichloride. Heating mixtures of barium, potassium, sodium amidosulphates or sulphamic acid, with sodium or potassium nitrates or nitrites, leads to reactions which may be explosive. Mixing sulphamic acid with fuming nitric acid results in violent release of nitrous oxide.

<b>Section 11. Toxicological Information</b>	
<b>Component Toxicity</b>	<b>LD50 Oral      LD50 Dermal      LC50 Inhalation</b>
Sulphamic Acid	1.45g/kg (Rat)
<b>Likely Routes of Exposure</b>	
<b>Skin:</b>	May cause redness and, blistering of skin, can lead to dermatitis.
<b>Eyes:</b>	May cause permanent damage to the eyes. Symptoms may include: irritation, burning, pain, watering and/or change of vision.
<b>Inhalation:</b>	Symptoms of exposure may include: nasal discharge, hoarseness, coughing, chest pain and breathing difficulty, general irritation of the respiratory tract. Accumulation of fluid in the lungs may occur (pulmonary oedema).
<b>Ingestion:</b>	May cause gastrointestinal irritation, nausea, vomiting, and diarrhoea. May cause liver and kidney damage.
<b>Acute Toxicity Estimates (ATE)</b>	Oral (Rat): 3.16 g/kg
<b>STOT (Specific Target Organ Toxicity) – Single Exposure</b>	Not classified
<b>Aspiration Toxicity</b>	Not classified
<b>STOT (Specific Target Organ Toxicity) – Repeated Exposure</b>	Not classified
<b>Skin Corrosion / Irritation</b>	Mild irritant
<b>Serious Eye Damage / Irritation</b>	Moderate irritant
<b>Respiratory or Skin Sensitization</b>	Not classified
<b>Carcinogenicity</b>	Not listed.
<b>Reproductive Toxicity</b>	
- <b>Sexual Function and Fertility</b>	Not classified
- <b>Development of Offspring</b>	Not classified

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- <b>Effects on or via Lactation</b>	Not classified
<b>Germ Cell Mutagenicity</b>	Not classified
<b>Interactive Effects</b>	Not classified
<b>Other Information</b>	None known

<b>Section 12. Ecological Information</b>	
<b>Ecotoxicity</b>	LC50: 14.2 mg/L (Pimephales promelas) 96 hr
<b>Persistence and Degradability</b>	Will not persist
<b>Bioacumulative Potential</b>	No data
<b>Biodegradability</b>	No data
<b>Mobility in Soil</b>	Not available
<b>Other Adverse Effects</b>	None known

<b>Section 13. Disposal Considerations</b>	
<b>Disposal Considerations</b>	Dispose of contents/container in accordance with local regulations.

<b>Section 14. Transport Information</b>	
<b>UN Number</b>	2967
<b>UN Proper Shipping Name</b>	Sulphamic Acid
<b>Transport Hazard Class(es)</b>	8
<b>Packaging Group</b>	III
<b>Environmental Hazards</b>	Not applicable
<b>Bulk Transport</b>	Not applicable
<b>Special Precaution</b>	Not applicable
<b>DOT Erg#</b>	154

<b>Section 15. Regulatory Information</b>	
<b>Canada – DSL Inventory</b>	All components of this product are either on the Domestic Substances List (DSL), Non-Domestic Substances List (NDSL), or exempt
<b>TSCA</b>	All components of this product are either on the Toxic Substances Control Act (TSCA) Inventory List or exempt
<b>Additional Information</b>	None

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### Section 16. Other Information

<b>NFPA Rating</b>	Health-2/ Flammability-0/Reactivity-2/Special Hazard-Not applicable
<b>HMIS Rating</b>	Health-2/Flammability-0/Reactivity-2/Personal Protection-See Section 8.
<b>Prepared by:</b>	Chemfax Products Ltd., Technical Department
<b>Date Prepared:</b>	28 September , 2012
<b>Date of Latest Revision:</b>	25 January 2021

#### Disclaimer

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